

## A TRITERPENE FROM *AKEBIA QUINATA* CALLUS TISSUE

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**Key Word Index**—*Akebia quinata*; Lardizabalaceae; callus tissue; triterpene; 3 $\alpha$ ,24-dihydroxy-30-norolean-12,20(29)-dien oic-28-acid; quinatic acid.

**Abstract**—A new triterpene was isolated from the methanol extracts of *Akebia quinata* tissue culture. The structure was determined by spectra and chemical transformations to be 3 $\alpha$ ,24-dihydroxy-30-norolean-12,20(29)-dien-28-oic acid.

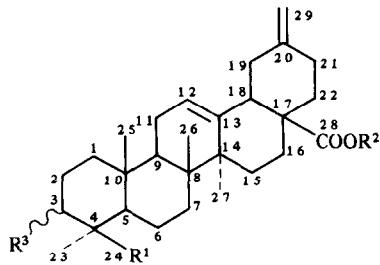
### INTRODUCTION

Recently we reported the isolation and structure determination of new triterpenes akebonic acid and its epimer together with two known triterpenes from the extract of the callus tissues of *Akebia quinata* (Lardizabalaceae, Japanese name Akebi [1]). In this paper, we report the isolation of a new triterpene.

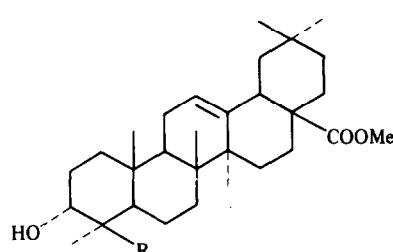
### RESULTS AND DISCUSSION

The crude triterpene mixture was obtained from the tissue culture as described in the Experimental. The mixture was chromatographed on HPLC repeatedly and compound **1** was obtained together with the compounds reported previously. The  $^1\text{H}$  NMR spectrum (pyridine- $d_5$ ) of **1** showed four tertiary methyl signals, two proton signals at  $\delta$  4.74(s) and 4.79(s) ascribable to the exomethylene protons and the hydroxymethylene protons at  $\delta$  3.83(d,  $J$  = 10.8 Hz) and 4.07(d,  $J$  = 10.8 Hz) which were shifted downfield on acetylation (**1b**) to indicate the presence of an axial hydroxy methylene group attached to an asymmetric centre. Moreover, the  $^{13}\text{C}$  NMR chemical shifts of the carbons of the C/D/E rings of **1** and akebonic acid (**2**) [1] are in good agreement with each other (Table 1). The mass spectrum of **1** showed  $[\text{M}]^+$  at  $m/z$  456 and exhibited a significant peak at  $m/z$  232 and 187, which could be assigned to the fragments of the D/E rings derived by retro-Diels-Alder cleavage of the  $\beta$ -amyrin- $\Delta^{12}$  skeleton [2]. From the above mass spectral fragments, it is also evident that the two hydroxyl groups are

present in the A/B ring portion. The axial ( $\alpha$ )-orientation of the secondary hydroxyl groups was confirmed by the  $^1\text{H}$  NMR data of **1** and **1b**. The  $^1\text{H}$  NMR spectrum of **1** showed the presence of a triplet-like signal centred at  $\delta$  3.70 and the signal shifted downfield to  $\delta$  4.92 on acetylation. Furthermore, the  $^{13}\text{C}$  NMR spectrum of **1** exhibited a signal at  $\delta$  70.2(d) for C-3 which was at higher field than that observed for related compounds with the 3 $\beta$ -hydroxy configuration [3]. From the above  $^1\text{H}$  NMR and  $^{13}\text{C}$  NMR spectral data a 3 $\alpha$ -hydroxyl group was presumed to be present. Furthermore, the location of the second hydroxyl group at C-24 was suggested, because **1a** did not form the acetonide between the C-3 $\alpha$ -hydroxyl and the 4 $\beta$ -CH<sub>2</sub>OH group on treatment by the usual procedure [4]. Other evidence for the C-4 stereochemistry was obtained by the comparison of the average chemical shift value of the acetoxy-methylene protons of **1b** [ $(a + b)1/2 = \delta$  4.06] with the  $^1\text{H}$  NMR data reported for similar compounds [5, 6]. Also the  $^{13}\text{C}$  NMR spectrum of the A/B ring of **1** similar to that of **3** [7] which was isolated from *Salvia nicolsoniana* or barbinervic acid which was isolated from leaves of *Clethra barbinervis* [8] (Table 1). The finding was further confirmed by correlating the signals due to the methyl protons of C-25 with those due to the methylene protons of the (C-24) group in the 2D NOE spectrum of compound **1a**. Thus, **1** was established as 3 $\alpha$ ,24-dihydroxy-30-norolean-12,20(29)-dien-28-oic acid (**1**) (quinatic acid). It is of interest that *A. quinata* callus produced the 30-noroleane type of triterpenoids.



	R <sup>1</sup>	R <sup>2</sup>	R <sup>3</sup>
<b>1</b>	CH <sub>2</sub> OH	H	---OH
<b>1a</b>	CH <sub>2</sub> OH	Me	---OH
<b>1b</b>	CH <sub>2</sub> OAc	H	---OAc
<b>1c</b>	CH <sub>2</sub> OAc	H	---OH
<b>2</b>	Me	H	■ OH



**3** R = CH<sub>2</sub>OH

Table 1.  $^{13}\text{C}$  NMR chemical shifts of compounds 1-3

C	Chemical shift			C	Chemical shift		
	1	2	3		1	2	3
1	33.7	39.0	33.9	16	24.0	23.8	23.5
2	26.4	28.1	26.4	17	47.1	47.1	47.0
3	70.2	78.2	70.0	18	47.9	48.0	41.9
4	43.9	39.4	43.9	19	42.0	42.0	46.2
5	50.1	55.9	50.1	20	149.2	148.5	30.8
6	19.1	18.8	19.1	21	38.4	38.4	34.0
7	33.7	33.3	33.6	22	30.4	30.4	32.8
8	40.0	39.8	39.9	23	23.6	28.2	23.4
9	48.1	48.1	48.1	24	65.8	16.5	65.8
10	37.5	37.4	37.5	25	15.9	15.6	15.9
11	23.8	23.8	24.0	26	17.3	17.4	17.1
12	123.0	123.3	120.0	27	26.1	26.2	26.1
13	144.2	143.5	144.1	28	179.4	177.3	177.9
14	42.2	42.1	42.0	29	107.1	107.1	33.1
15	28.3	28.3	28.1	30			23.7
					COOMe		51.5

All signals were corroborated by DEPT techniques. The measurements were made in pyridine- $d_5$  with TMS as an internal standard.

## EXPERIMENTAL

Mps: uncorr. The  $^1\text{H}$  NMR spectra were recorded at 400 MHz and the  $^{13}\text{C}$  NMR spectra were recorded at 100.6 MHz, at room temp. with  $\text{CDCl}_3$  solns and pyridine- $d_5$  soln and TMS as int. standard. MS (70 eV) were taken with a direct probe.

*Plant material.* *Akebia quinata* Decne was collected in October, 1981 at the Medicinal Plant Garden of this college.

*Derivation and culture of callus tissue.* The callus tissue from the stalk was obtained in October, 1981. Murashige and Skoog's medium containing 2,4-D (1 mg/l; 3 mg/l) and kinetin (0.1 mg/l) as plant growth regulators were used for induction of callus tissue. The callus tissue was subcultured every 5~6 weeks onto fresh M & S medium (minus glycine) containing 2,4-D (1 mg/l) and kinetin (0.1 mg/l) at  $26^\circ \pm 1$  in the dark.

*Extraction and isolation.* The fresh callus tissue (1092 g, fr.wt, dry, wt 22.4 g) was extracted with cold MeOH and EtOAc in a Waring blender. The extracts were combined and concd under red. pres. to yield an extract which was partitioned between  $\text{CHCl}_3$  and  $\text{H}_2\text{O}$  to obtain the organic solvent soluble fraction and the residual  $\text{H}_2\text{O}$  soln was further partitioned with *n*-BuOH saturated with  $\text{H}_2\text{O}$ . The BuOH soln was chromatographed over a column of silica gel (Merck 9385) and elution with  $\text{CHCl}_3$  containing increasing proportions of MeOH afforded the crude triterpene mixture. The mixture was purified by repeated re-chromatography on a silica gel column [HPLC, CIG column system (Kusano Scientific Co., Tokyo) with Iatrobeads as the stationary phase (60 silica gel, IATRON Co., Tokyo)] to afford compound 1.

*3 $\alpha$ ,24-Dihydroxy-30-norolean-12,20(29)-dien-28-oic acid (1) (quinatic acid).* Mp 269~272°, colourless needles  $[\alpha]_D^{25} + 66.6^\circ$  (pyridine;  $c$  0.375).  $^1\text{H}$  NMR (pyridine- $d_5$ ):  $\delta$  0.96 (3H, s), 1.01 (3H, s), 1.16 (3H, s), 1.62 (3H, s), 3.22 (1H, dd,  $J = 4, 12$  Hz), 3.83 (1H, d,  $J = 10.8$  Hz), 4.07 (1H, d,  $J = 10.8$  Hz), 4.43 (1H, br s), 4.74 (1H, s), 4.78 (1H, s), 5.51 (1H, br s). MS  $m/z$  rel. int.: 456 [ $\text{M}]^+$  (2), 232 (100), 224 (21), 206 (47), 187 (97), 175 (52).

*Methyl-3 $\alpha$ ,24-dihydroxy-30-norolean-12,20(29)-dien-28-oate (1a) (quinatic acid methylester).* Mp 233~236° (MeOH- $\text{CHCl}_3$ ), colourless needles.  $[\alpha]_D^{25} + 110.8^\circ$  ( $\text{CHCl}_3$ ;  $c$  0.318), IR  $\nu_{\text{max}}^{\text{CHCl}_3}$   $\text{cm}^{-1}$ : 3600, 1710.  $^1\text{H}$  NMR ( $\text{CDCl}_3$ ):  $\delta$  0.70 (s, 3H), 0.89 (s, 3H), 1.08 (s, 3H), 1.17 (s, 3H), 2.73 (dd,  $J = 4.0, 13$  Hz), 3.48 (1H, d,  $J = 11$  Hz), 3.67 (1H, d,  $J = 11$  Hz), 3.60 (s, 3H), 3.82 (1H, t,  $J$

$= 3$  Hz), 4.61 (1H, s), 4.63 (1H, s), 5.34 (1H, t,  $J = 3$  Hz) MS  $m/z$  (rel. int.): 470 [ $\text{M}]^+$  (6), 452 (2), 411 (4), 393 (2), 246 (55), 206 (25), 187 (100), 186 (65).

*Acetylation of 1.* Compound 1 was treated with  $\text{Ac}_2\text{O}$  and pyridine at room temp. over night. Ice was added to the reaction mixture, which yield a white ppt. The ppt. was filtered, dried and purified by HPLC. It afforded two colourless amorphous crystals (1b, diacetate) and (1c, monoacetate).

**1b (diacetate):**  $^1\text{H}$  NMR ( $\text{CDCl}_3$ ):  $\delta$  0.73 (3H, s), 0.95 (6H, s), 1.22 (3H, s), 2.03 (3H, s), 2.09 (3H, s), 2.49 (1H, t,  $J = 13.5$  Hz), 2.75 (1H, dd,  $J = 13.1, 4.5$  Hz), 3.95 (1H, d,  $J = 11.4$  Hz), 4.17 (1H, d,  $J = 11.4$  Hz), 4.64 (2H, s), 4.97 (1H, t-like), 5.35 (1H, t-like). MS  $m/z$  (rel. int.): 540 [ $\text{M}]^+$  (5.2), 480 (7.9), 420 (5.7), 233 (17), 232 (84), 204 (17), 203 (7.6), 189 (25), 188 (59), 187 (100). **1c (monoacetate):**  $^1\text{H}$  NMR ( $\text{CDCl}_3$ ):  $\delta$  0.75 (3H, s), 0.93 (3H, s), 1.06 (3H, s), 1.19 (3H, s), 2.04 (3H, s), 2.51 (1H, t,  $J = 13.8$  Hz), 2.78 (1H, dd,  $J = 13.1, 4.7$  Hz), 3.74 (1H, t-like), 3.97 (1H, d,  $J = 11.4$  Hz), 4.19 (1H, d,  $J = 11.4$  Hz), 4.64 (2H, s), 5.36 (1H, t,  $J = 0.3$  Hz). MS  $m/z$  (rel. int.): 498 [ $\text{M}]^+$  (2), 480 (3), 452 (2), 420 (2.4), 232 (78), 204 (13), 203 (20), 189 (22), 188 (65), 187 (100).

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